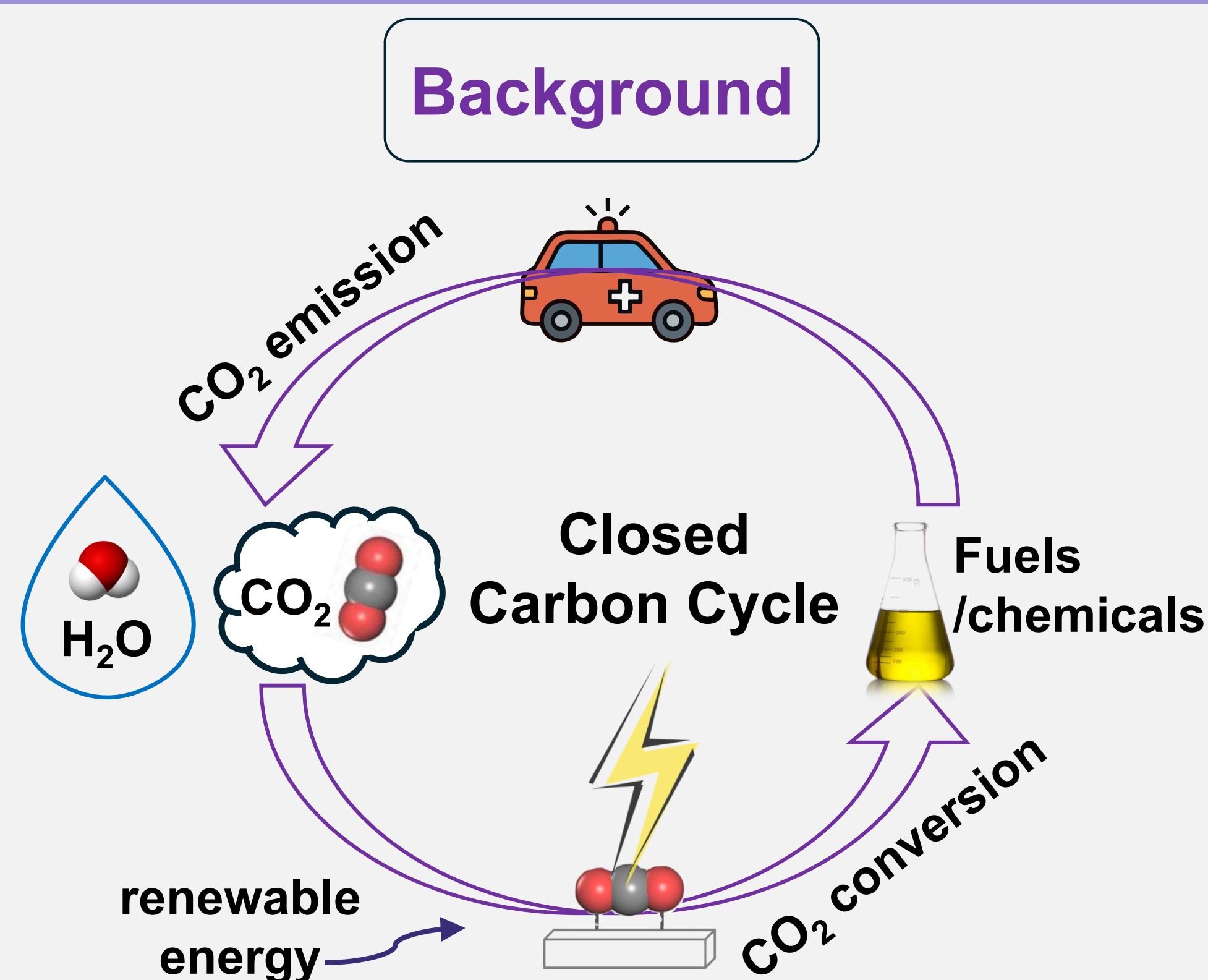


Introduction

Background



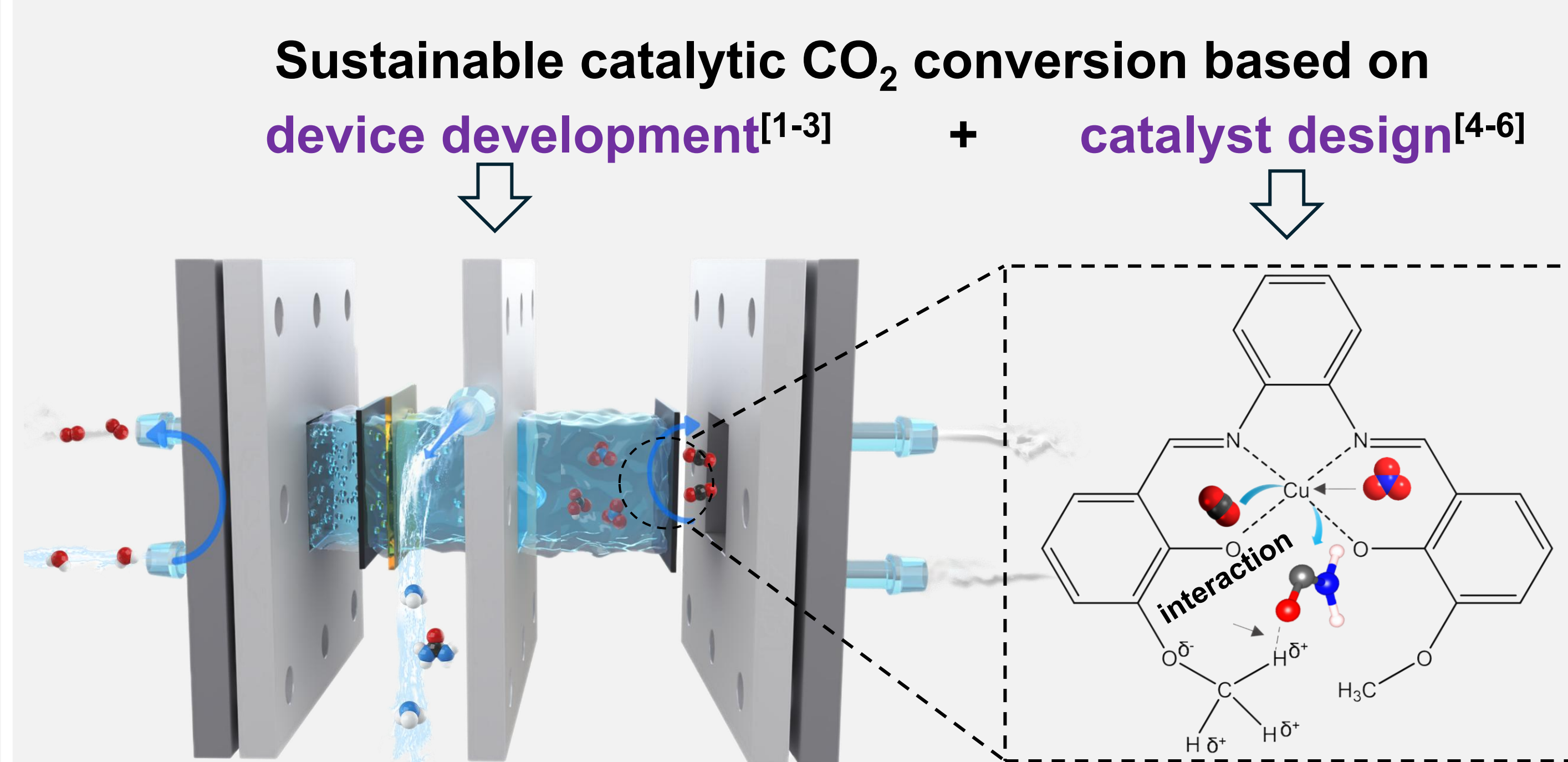
Motivation

- To convert exiting CO₂ emissions back to valuable chemicals
- To reduce high CO₂ emissions during industrial manufacturing of urea
- To transform chemical industrial manufacturing in a sustainable way

Challenge

- Unclear mechanism during reactions
- Low efficiency of catalysts
- Low production rate in reactors

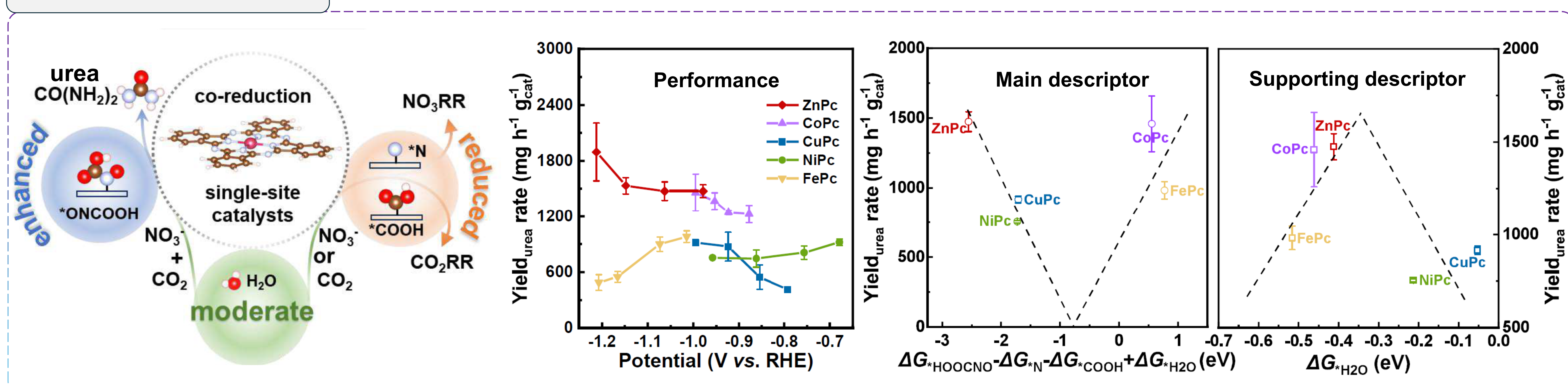
Methodology



Key Findings

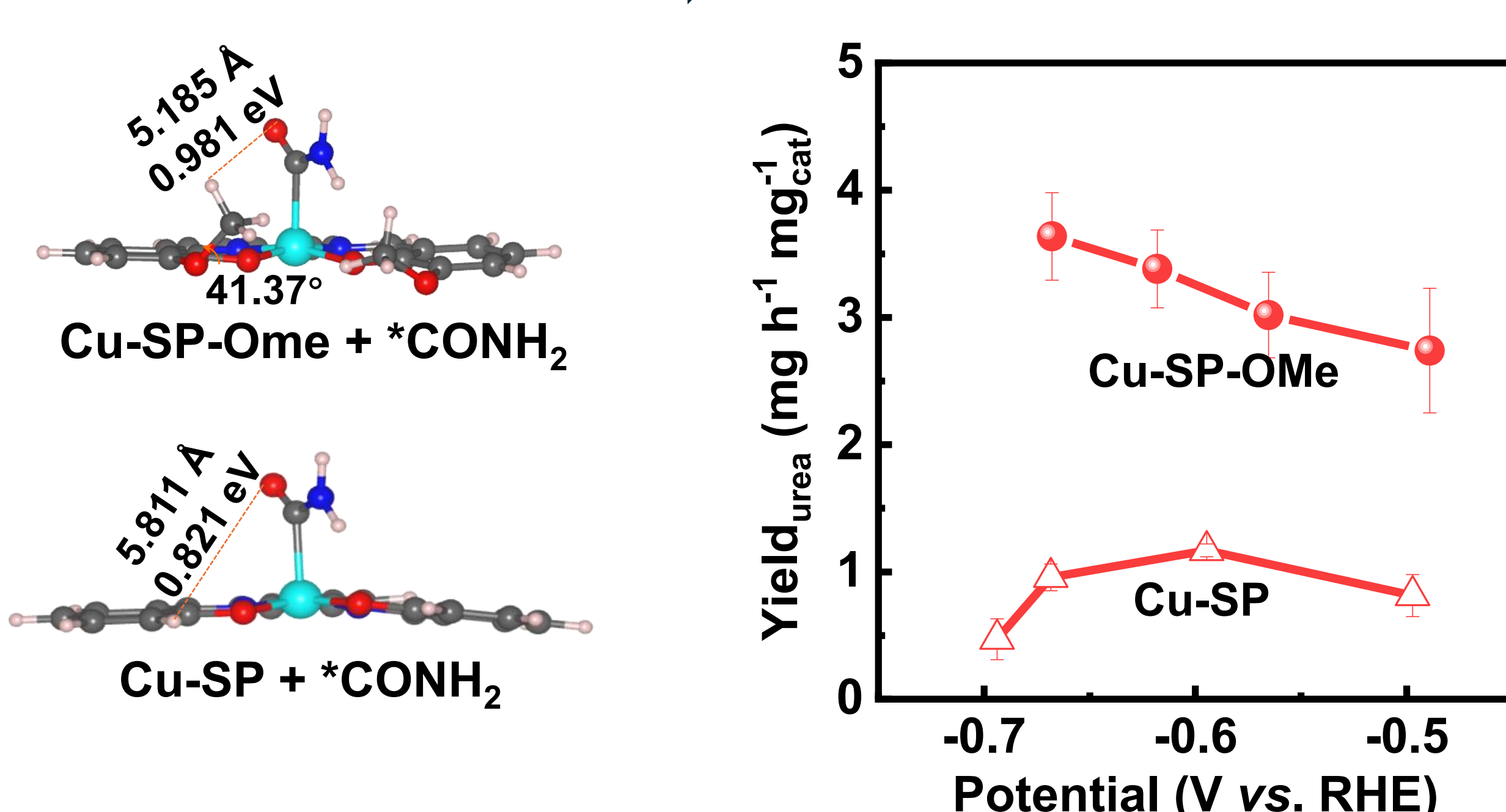
1. Mechanistic Exploration

➤ $\Delta G_{*HOOCNO} - \Delta G_{*N} - \Delta G_{*COOH} + \Delta G_{*H_2O}$ can be a potential descriptor for urea formation.^[5]



2. Catalyst Design

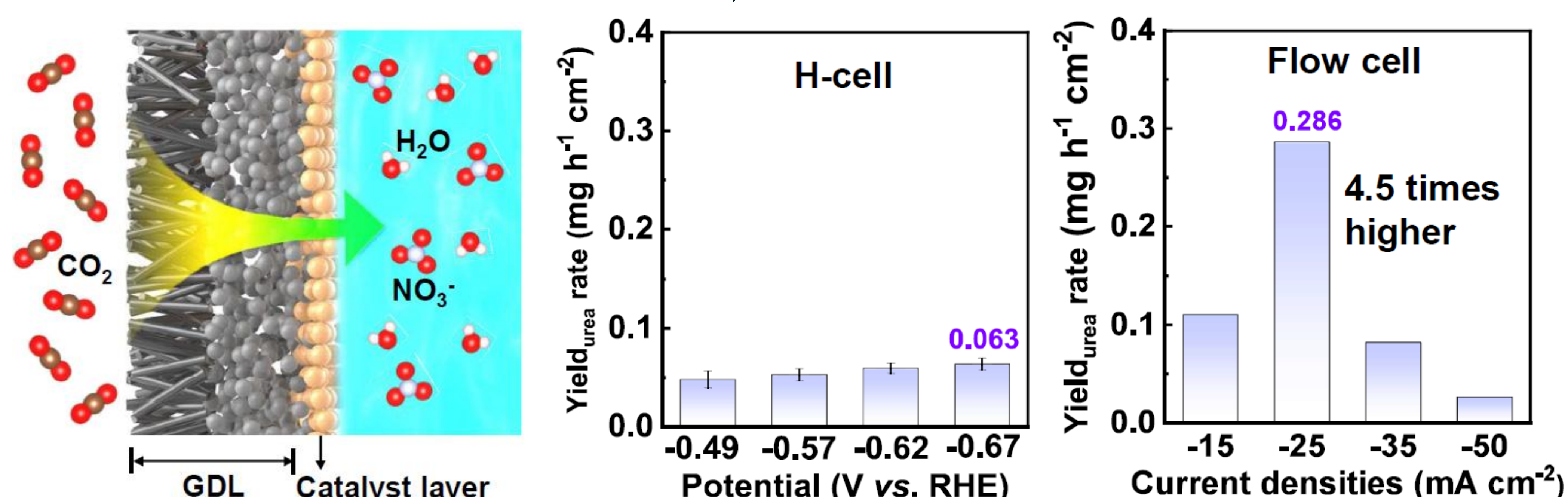
Optimized binding ➡ Increased yields on catalysts



➤ The substituted -OCH₃ group helps binding *CONH₂ during C-N coupling process toward urea formation.^[2]

3. Device Development

Three-phase interface engineering ➡ Improved production in flow cell



➤ The creation of gas-solid-liquid three-phase interface^[1] eliminates mass transport issue caused by low solubility of CO₂,^[2] promoting reaction process.

Summary and Perspectives

- CO₂-involved electrocatalytic conversion offers a highly **sustainable pathway** for producing value-added chemicals using renewable electricity.
- **High-activity and high-selectivity catalysts** are the key to achieving high-performance conversion to desired products.
- **Device development** is critical for overcoming mass transport limitations at reaction interface and achieving industrially relevant production rates.
- Descriptors can be integrated with **artificial intelligence** to guide rational design and accelerate screening of potential catalysts.

References

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 [3] Q. Zhao, et al., *Nat Water*, 2026, under review.

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 [5] Q. Zhao, et al., *Adv. Sci.*, 2025, 2501882.
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Organizers

